



The Use of Various Polymers with Solid Dispersion Application of Hot Melt Extrusion (HME) Preparing New Delivery Drug System

Sofi Nurmay Stiani^{1, 3*}, Taofik Rusdiana¹, Anas Subarnas²

¹ Department of Pharmaceutics and Pharmaceutical Technology, Universitas Padjadjaran, Indonesia.

² Department of Pharmacology and Clinical Pharmacy, Universitas Padjadjaran, Indonesia.

³ STIKes Salsabila Serang Banten, Indonesia.

*Corresponding Author: Sofi Nurmay Stiani

Abstract

Hot Melt Extrusion (HME) at the beginning appeared widely used in the plastic industry and food industry, but in the last few decades, many researchers have succeeded in applying the HME method as a drug delivery system and determining various dosage forms in the pharmaceutical industry. The challenges that arise in the pharmaceutical are related to the solubility, dissolution rate, and bioavailability of drugs, especially for drugs that have low solubility in water. The low solubility and bioavailability will reduce the pharmacological effect of a drug. Solid dispersion using HME method succeeded in answering this challenge by distributing the drug evenly in the hydrophilic polymer during the extrusion process and changing the crystalline form to the amorphous form while maintaining drug stability. This review will examine the polymers are suitable for the HME method, Mechanism for solid dispersion using HME method, Steps must be taken to prepare solid dispersions using HME and characterization of solid dispersion resulting from HME and determinants of success of solid dispersions

Keywords: *Polymer, Solid Dispersion, hot-melt extrusion, Mechanism of solid dispersion, Step HME.*

Introduction

Hot Melt Extrusion (HME) in the pharmaceutical world was introduced in the 1970s [1], this method is used in the pharmaceutical world to prepare solid dispersions. HME processing by mixing API (Active Pharmaceutical Ingredient), polymer, and if necessary plasticizers, surfactants, or others can be added to the screw extruder by adjusting the temperature and extruder rotation so that the API dissolves in amorphous polymers because there is a melting process, mixing, and homogenous, at the end it comes out through the die and there is a cooling and pelletizing process [2].

The HME method is a method that is more suitable than spray dry to prepare solid dispersions of drugs that have high crystallinity [3]. HME is a promising technology in the pharmaceutical industry,

because it will not cause a problem with the toxicity of the residual solvent, does not require a solvent in the manufacture of solid dispersions. Besides, this method is also environmentally friendly [4], reduces process stages, does not require compressibility of active substances [5] and has been shown to disperse drug molecules with difficult solubility in polymer-carriers as well as to increase dissolution rates and bioavailability [6].

Several researchers have provided numerous reports on the use of polymers to disperse the active substances molecularly, help stabilize solid preparations, prevent crystallization, and maintain supersaturation during dissolution [7]. The following are some studies on the use of polymers in the pharmaceutical field in the last 10 years:

Table 2: Use of HME Application Polymers

No	Year	Active Pharmaceutical Ingredient	Polymer and plasticizer	Drug to polymer ratio	Tg (°C)	Tm(°C)	Dosage Form	Result	Ref
1.	2013	Carbamazepine	Soluplus	1:9 2,5:7,5 3:7 5:5	75,6°C soluplus	193,7°C CBZ in soluplus	Extrudate	Crystallinity is reduced at 15% w / w CBZ compared to 25% immiscibility occurs at 30% CBZ and 50% w / w	[8]
2	2019	Glyburide	Soluplus Kollidon VA 64 added PEG 4000 as a plasticizer	1:1 1:2	107°C kollidon	175°C glyburide	Extrudate	More molecularly dispersed in soluplus polymer than PVP VA 64, however, X-ray microtomography showed an increase in polymer, decreased porosity of HME dispersion with PVP VA 64 potentially makes the product solid.	[9]; [10]
3.	2018	Meloxicam (MLX)	Kollidon VA 64 with meglumine (MGL) and Eudragit® EPO as a plasticizer	1:0 MLX:MGL 0,5:1 MLX:MGL 1:1 MLX:MGL 3:1 MLX:MGL 1:1 MLX:EPO 3:1 MLX:EPO 5:1 MLX:EPO	106 °C Kollidon; MLX 17 °C and meglumine 18 °C	-	Extrudate	Meloxicam concentration on 10% (w / w) with the same amount of melumin which has the optimum result	[11]
4.	2014	Osthole Derivative coumarine	Plasdone S-630, HPMC-E5, Eudragit EPO, and Soluplus	1:3 1:6 1:9	1:3 SD Soluplus 75,9 °C Eudragit EPO 77,6°C HPMC 79,6 °C	Osthole 86,5 °C	Extrudate	Plasdone S-630 or HPMC-E5 (drug/polymer: 1:6) significantly increased the dissolution rate (3 fold D30)	[12]
5.	2018	Itraconazole	Kollidon®V A64 with acid glutaric as an acid for pH decreases and solubility increases	1:4:1 Itra:acid:koli	90°C	Itracozazole 168°C; acid glutaric 98°C	Fast Dissolving Tablets	Itraconazole solid dispersion is physically stable in combination acid glutaric	[13]
6.	2019	Acetaminophen or APAP (amorphous N-acetyl-para-aminopheno	HPMC	30:70 b/b	-	Acetaminophen 169–170°C	Extrudate	APAP solid dispersion with HPMC is best in high spatial	[14]

		l)						resolution (~ μm) capture images at 1700C and 200rpm	
7.	2018	Itraconazole	Copovidone	1:1 1:3 1:4 dan 1:9	-	Itraconazole 166°C	Extru date	The dissolution test results of SD Itraconazole increased 18 times compared to the original Itraconazole	[15]
8	2019	Indomethacin (IND) and Fenofibrate (FEN)	Soluplus	1:4	Around 80°C	Indomethacin 60,43°C Fenofibrate 82,66 °C	self-micellizing solid dispersions	Selfmicellizing Soluplus carrier is capable of dissolving IND or FEN and suppressing drug crystallization from a saturated state.	[16]
9.	2017	Indomethacin (IMC)	Soluplus and Mannitol as an excipient	50:50 15:85 IMC:SOL:MAN 50:50:25 IMC:SOL:MAN 15:85:15	SOL 71,1°C IMC 42°C MAN 13°C	IMC 165°C	Extru date	IMC extrudate was stable over 2 weeks	[17]
10	2019	Fenofibrate (FNB)	PVP VA64	2:8	FNB °C 85,5 PVP VA64 106°C	-	Extru date	Compared with the commercial lypanthil product, the bioavailability of the HME solid dispersion increased 2.45fold	[18]
11	2015	Artemether (ARTM) and lumefantrine (LUMF)	Soluplus added surfactant PEG 400, LF 127, LF 68	1:1 1:2 1:3	89,8 °C 70,7°C 67,2°C	-	Extru date	Solubility test, dissolution significantly increased compared to pure drug and pharmacokinetic studies SD AL1 (sol: PEG4000) 1: 1, namely 44.12–65.24 fold increase AUC (0–72) and 42.87–172.61 fold increase Cmax	[19]
12	2018	Indomethacin (IMC)	Partially hydrolyzed polyvinyl alcohol (PVOH)	3:7 5:5 7:3	Tg PVOH 60,8°C Tg PVOH HME (170°C)=	Tm PVOH 150 - 200°C	Extru date	HME succeeded in amorphizing IMC with	[20]

					59,8°C Tg PVOH HME (190°C)=60, 4°C Tg γ-IMC = 160,3°C			the inter- molecular interaction between the IMC carboxylic acid and the hydroxy PVOH group and resulted in a stable solid dispersion.	
13	2020	Curcumin	Eudragit RSPO(hypot onicity) and Eudragit RLPO(hyper tonicity)	1:3 1:6 with a certain amount of porogen for sustained release	Tg curcumin 170°C	-	Sustaine d release solid dispersio n	The effect of solid dispersion curcumin is significant when compared to physical mixtures and pure curcumin.	[21]
14	2018	Tamoxifen dan Resveratrol	Soluplus, Eudagit EPO, dan Kollidon VA64	1:2 1:5	Tg 144 °C Eudragit EPO and Kollidon VA64 at D:P ratio 1:2	Tm tam = 266°C	Film	HME significantly increased the bioavailabi lity of tamoxifen	[22]
15	2018	Lacidipine	Soluplus, PVP K30, and PVP VA64)	1:5 1:10 1:15	Soluplus (Tg 70°C), PVP VA64 (Tg 101°C), and PVP K30 (Tg 149°C) Tg SD 66°C soluplus, Tg SD 81°C PVP VA64	Tm= lacidipine = 183°C	Extru date	The optimum result is 1:10 with soluplus polymer and PVP VA64	[23]
16	2019	Itraconazole	HPMCAS and surfactant	Polymer: drug:surfactant 70:20:10 dan 65:20:15	Tg film contain-ing 30% P407 155°C	Tm itra Cona zole = 165°C	Extru date	HME improves drug stability and release	[24]
17	2014	Sulindac	SUL:PPP:P EG1000 SUL:HPMC: TEC SUL:HPC:P EG1000	drug:polymer: plasticizer 1:5:1	Tg PPP 70°C	Tm Sulindac 188°C	Film	The combinatio n of SUL + PPP + PEG1000 which provides the best and stable results	[25]
18	2017	Artemisinin	Soluplus and acid citric as an acidifier	1:1 1:0,95:0,5 Drug:polymer: citric acid	Tg soluplus 70°C	Tm artemisinin 154,81 °C	Extru date	SD stabilized at 12 months	[26]
19	2019	Aripirazole (ARI)	Kollidon® 12 PF (PVP) and succinic acid (SA)	Drug:polymer:a cidifier 10:70:10 40:50:10 30:65:5	-	Tm ARI 140°C Tm SA 190°C	Tablet	HME results were able to increase the solubility, dissolution and oral bioavai- lability	[27]
20	2019	Tacrolimus	Polyvinylpyr rolidone vinyl acetate (PVP VA64), Soluplus®	10:10	Tg TAC= 78,8°C Tg PVPVA64= 101°C	Tm TAC above 130°C	oral Disintegr ation tablet	ODT tablets were stable for 3 months	[28]

			and Hydroxypropyl Cellulose (HPC), at a drug loading of 10% w/w		Tg Soluplus=70 °C Tg HPC=105°C			and increased their dissolution rate	
21	2017	Felodipine	Hydroxypropyl cellulose (Kluce TM MF) and hypromellose (Benece TM K15M)	1:1	-	Tm Felodipine 146°C	Bioadhesive floating pellets	pellets can serve as a platform for creating gastroretentive controlled-release DDS	[29]
22	2018	ibuprofen (IBU) and isonicotinamide (INA)	xylitol, PVP K15 Soluplus	Drug:polymer 90:10	Tg polymer Xylitol 84,95 and Tg co crystal 115, 28°C Tg soluplus dan Tg co crystal 118, 30°C 65,81 Tg PVP k-15 139,22 °C dan Tg co crystal 114, 31°C	-	Co crystal	spray drying showed better than HME in making Co Crystal without excipient carrier.	[30]
23	2019	Mefenamic acid	Eudragit® EPO	Drug:polymer 1:3 and 1:4	-	Tm Mefenamic acid 230 °C	Tablet	Mefenamic acid tablets were successful with Eudragit® EPO polymer to increase the solubility of materials with high melting point and low solubility.	[31]
24	2015	Valsartan (VST)	Soluplus® (SP) and d-alpha-tocopherol polyethylene glycol 1000 succinate (TPGS) as a plasticizer	Polymer:drug 9:1, 7:3, and 5:5 VST:SP 30%:70% VST:SP:TPGS 30%:60%:10%	Tg soluplus 70°C	Tm TPGS 40°C; Tm VST 104,6°C	Extrudate	Formula developed 7:3; the addition of a plasticizer decreases Tg and the melting viscosity of polymers during extrusion processing; HME successfully improves oral bioavailability	[32]
25	2017	Gliclazide (GLZ)	Affinisol, Soluplus®, Kollidon VA64®	10:90	-	-	Extrudate	Miscibility of affinisol is better than Soluplus®, Kollidon VA64® and degradation does not	[33]

								depend on the polymer	
26	2016	Oleanolic acid (OA)	Soluplus®, PVP VA64 and PEG 6000	1:3; 1:5; 1:7; 1:10; 1:15	-	Tm OA 320°C	Ekstru date	PVP VA64 with 1:10 which was amorphous and homogeneous and increased bioavailability	[34]
27	2018	Efavirenz (EFV)	Soluplus® (SOL) and Kollidon® VA 64 (KVA64)	70:30; 50:50; 30:70	Tg EV 139.60 °C	-	tablet	The maximum solubility and dissolution rates are in the 30% EFV formula with both SOL and KVA64 alone.	[35]
28	2016	Felodipine (FEL)	Soluplus® (SOL)	1:1; 1:9;3:7	Tg SOL 70°C	-	Extru date	The dissolution is increased through solid amorphous dispersions and encapsulation of SOL micelles	[36]
29	2015	Felodipine (FEL)	PEG 4000; polyox WSR 1105; Tween 80	FEL:PEG:POL: TWEEN 10% 36% 27%27% 20% 32% 24%24% 30% 28% 21%21%	Tg FEL 45°C Tg PEG - 61°C; Tg PEO-56°C; Tg tween - 65°C	Tm FEL 144,39°C Tm PEG 59,13°C; Tm PEO 70,64°C	Buccal Patches	FEL Crystals have good solubility in tween 80	[37]
30	2017	Naproxen	Meglumine (acidifier) Soluplus Kollidon VA 64 PVPK30	NPX:Meg:polymer 48.7:41.3:10.0 48.7:51.3:31.63 48.7:41.3:10.0 48.7:51.3:43.70 48.7:41.3:10.0 48.7:51.3:49.00	Tg Nap 5°C Tg meg 17°C Tg Kollidon K30 149°C	Tm Nap 155°C Tm meglumin 129°C	Extru date	The acid-base reaction between NPX and MEG during melt extrusion significantly improved the physical stability and the dissolution rate of NPX ASDs.	[38]
31	2016	Ibuprofen (IBU)	Eudragit RS PO (EU) EU/Sucrose EU/Methylcellulose EU/Xanthan gum EU/Poloxamer EU/Gelucire 44/14	30%:70% 60:10 60:10 & 10:60 60:10 & 10:60 60:10 60:10	Tg Ibu - 43,4°C; Eudragit 53,3°C; metil selulosa 103°	Tm ibu 75,7°C; sukrosa Tm 175,2°C; Gelucire39,5°C	Transdermal Film	The best film on Gelucire 44/14 polymer	[39]
32	2012	Bifendate (BFN)	Plasdone ® S-630 Eudragit ®	10:90 20:80	-	Tm BFN 179,16°C; Tm PEG 6000	Extru date	Kollidon® VA 64 is the optimal	[5]

			EPO Kollidon® VA 64	10:90		63,12°C		polymer in HME bifendate	
33	2016	Albendazole (ALB)	PVP K12	10:90 w/w	Tg PVP K12 90°C	ALB Tm 208°C; TM PVP K12 145°C	Extrudate	Significantly increased dissolution and was stable at 6 months of storage	[40]
34	2018	Phenytoin (PHT) Griseofulvin (GSF) Ibuprofen (IBU) Loratadine (LOR)	Poloxamer 338 (PLX) and dendrimer-like biopolymer (DLB)	API:DLB: PLX 3:1:1	-	Tm PHT 296,23°C GSF 216,86°C IBU 74,79°C, LOR 134,06°C	Nano Partikel Edible	Increases solubility and dissolution rate. Solubility in water PHT <GSF <MOTHER <LOR.	[41]
35	2015	Carbamazepine	Soluplus® and hypromellose acetate succinate (HPMCAS-HF)	API: HPM-HF:solu 20:8:72 20:64:16 20:24:56 20:32:48 20:40:40	-	Tm 191,01°C	Ekstrudate	Soluplus increases the dissolution rate, the addition of HPMCAS HF increases stability for up to 12 months due to its high Tg and low hygroscopic properties	[42]
36	2017	Mefenamic Acid (MA)	Kollidon 12PF and Kollidon 17PF, added MgO as alkalizer) and PEG 3350 as a plasticizer	20:80 20:70:10 (PEG) 20:75:5 (MgO) 20:65:5(MgO):10 (PEG)	Tg 12PF 90°C; Tg 17PF 138°C	Tm PEG53-59°C Tm MgO 2852°C Tm 230°C	Ekstrudate	Penambahan an alkalizer with plasticizer in the formula has a significant effect on the release of API from the kollidon matrix	[43]
37	2018	CuSO4	(CuSO4:Span 80:Tween 80:PEG 6000)	20:12:4:64	-	-	Nanocomposites (NCs) from CuSO4	CuSO4 NCs are successful photothermal candidates for colon cancer therapy	[44]
38	2018	Ketokonazole (KNZ)	PVP-VA64	40:60	Tg PVP-VA64 103,6°C Tg SD 69,3°C	Tm KNZ 149°C	Ekstrudate	The polymer aids in supersaturated and inhibits API deposition in the amorphous solid dispersion	[45]
40	2020	Quersetin	PEG 6000, F68, Soluplus	Drug-polymer(s) system w:w	-	Tm Que 326°C, HPMC 225°C, F68 is 57°C,	Ekstrudate	The best excipient F68	[46]

			and PVP VA64	Que:F68 1:7 Que:PEG 6000 1:7 Que:PVP VA64 1:7 Que:Soluplus 1:7 Que:F68 1:3 Que:F68 1:5 Que:F68 1:9 Que:F68 1:12 Que:F68:HPMC E5 1:6:1 Que:F68:HPMC E5 1:5:2 Que:F68:HPMC E5 1:4:3		PVPVA64 is 130°C, soluplus 70 °C and PEG 6000 57°C		increases dissolution	
41	2018	Ovalbumin (OVA)	Glycerol tristearate (D118) and hydrogenated palm oil (DP60)	-	Tg D118 70°C Tg DP60 50,02°C	Tm OVA 84.5 °C	Lipid based implant	Release from D118-based implants faster than the DP60implants	[47]
42	2019	Etravirine (ETR)	PEG , PVP , PVPVA,SLP , HPMCAS , HPMC	1:3	ETR 100,85°C PVP K12 90°C PVPVA 101 °C SLP 70°C HPMC 178°C HPMCAS–MG 130°C	Tm ETR 226°C	Ekstrudate	Drug release PEG > PVP > PVPVA > SLP > HPMCAS > HPMC	(48)
43	2018	Atovaquone for brain cancer	PVP K30+ Spontaneously Emulsifying Component (SEC)	API:polimer+SEC 20:70:5	Tg PVP K30= 172,81°C	Tm ATO 220,75°C	Extrudate	Incorporate in the SEC accelerate dissolution path	[49]
44	2017	Naproxen	Povidone K25	30%:70%	Tg Nap 7.8 °C Tg PVP K25 155°C	-	Extrudate	NPX 30% with HME results in a stable solid dispersion	[3]
45	2020	Ibrutinib (IBR)	HPMCAS , PVPVA ,PEG ,SLP, PVP , HPMC , HPC , PVOH	SLP + 50% IBR + 10% PEG6000 SLP + 50% IBR + HPMCAS. PVPVA + 50% IBR + 15% PVP k-12 + 50% IBR + 10% PEG6000	Tg IBR 79,1°C PVP K12 90 °C PVPVA 101 SLP 70°C HPMC 178°C HPMCAS - MG 130°C	Tm IB 152,2°C	Ekstrudate	Solubility HPMCAS > PVPVA > PEG > SLP > PVP > HPMC > HPC > PVOH	[50]
46	2011	Celexocib (CX)	Eudragit 4155F and polyvinylpyrrolidone (PVP)	drug/polymer ratios of 1:9, 3:7, 1:1 and 7:3	Tg PVP 154,6°C Tg SD PVP 3:7 130,70°C Tg SD PVP 1:1 14,37°C Tg SD PVP 88,4°C Tg (1:9) SD 51,30°C Tg(3:7) SD 52,70°C Tg (1:1) SD 55,10°C Tg (7:3) SD 59,20 °C	-	Ekstrudate	Eudragit 4155F significantly increases the solubility of CX	[51]
47	2020	Celexocib	Soluplus® With 120 mg SD, lactose 120 mg, micro-crystalline cellulose, 90	16,7:83,3 25:75 50:50 55:45 60:40 65:35 70:30	-	Tm CX 158°C	Tablet	Process analytical technology (PAT) for real time API analysis.	[52]

			mg, 30 mg of Kollidon VA 64, 32 mg sodium starch glycolate, and 8 mg magnesium	75:25 83,3: 16,7				The F4-F9 has increased its success	
48	2020	Diclofenac sodium	polyethylene glycol 400 added zein	12.5:10 12.5:20 18.75:20 25:20 37,5:20	Tg zein 168,4 °C	Tm 287,55 °C	Controlled release	The extrudate with zein-based diclofenac sodium was successfully produced	[53]
49.	2020	Glycyrrhetic Acid (GA)	Kollidon® VA64 and L-arginine/meglumine as alkalizers	1:1:8 1:2:7 1:3:6 2:3:5	-	Tm GA 305°C, Tm LA 228°C, and Tm MG 129°C	Extrudate	The addition of an alkalizer increases the dissolution of ionizable ones such as GA	[54]
50	2019	Aripiprazole (ARP)	Kollidon VA64; Soluplus	ARP/KVA 70 wt%, ARP/SOP 85 wt %	-	Tm ARP 395 K	Extrudate	Kollidon VA64 as the best polymer for amorphous Aripiprazole extraction	[55]
51	2018	Haloperidol	Kollidon® VA64+Affinisol™15 cP, Kollidon® VA64+HPM CAS,	drug-polymer-polymer (1:5:5 and 2:5:5)	Tg Affinisol™ 15 cP (98°C) and Kollidon® VA64 (108°C)	Tm 152°C	Tablet	1: 1 from Kollidon® VA64 and Affinisol™ 15 cP are polymers suitable for 3D printing and rapid drug release	[56]
52	2012	Bicalutamide (BL)	poly(ethylene oxide) (PEO)	1:10, 2:10 and 3:10	Tg BL 56,4°C	Tm BL 196°C	Ekstrudat	The solubility of BL in PEO liquid is estimated to be 31%	[57]
53	2015	Bicalutamide (BL)	Kollidon VA64	5% and 30% bicalutamide	-	-	Ekstrudat	Extrudate 30%, dissolution depends on the physical and chemical properties of BL	[58]
54	2016	Indomethacin (IMC)	Kollidon VA64	5%, 15%, 30%,50%,70%,90% drug loading	-	-	Ekstrudate control release	At pH 6.8 phosphate buffer the drug release increases in the drug extrudate 15% or more, at pH 2 HCl the drug release buffer increases at 5% drug	[59]

55	2020	Ibuprofen	cellaburate:colophony	CLB:CLP:Ibu 30:55:15 25:45:30 20:35:45	Tg polimer 74,3°C Tg Cellaburate 130°C	Tm film 157,2°C Tm Cellaburate 155-165 °C	Extrude film Control release	Ibuprofen is amorphous at 30% (w / w) in 35:65 colophony: cellaburate films.	[60]
56	2020	Indomethacin	Kollidon VA64	1:1	Tg Kollidon VA64= 105 °C	Tm 160 °C	Extrude	Solid dispersions contain residual crystallinity which affects the seed properties, polymeric crystal growth inhibition effectiveness, and supersaturation conditions.	[61]
57	2016	Piperine	Eudragit EPO, Kollidon VA 64, Soluplus	10:90 20:80 40:60	-	Tm piperine 135°C	Ekstrude	10% w / w piperine / Soluplus significantly increases solubility and dissolution	[62]
58	2020	Meloxicam	Soluplus®+ Poloxamer	API:SOL:POL 2.5%:2,3:1	-	-	Ekstrude	HME results have increased bioavailability compared to the innovator Mobic®	[63]

Discussion

Polymers in the Hot Melt Extrusion Method

The use of polymers in HME must meet the requirements, namely, thermoplastic has a glass temperature between 50-1800C, has high-temperature stability, is not toxic, has low hygroscopicity to prevent crystallization [64]. Some of the polymers that are often used for HME are as follows:

Eudragit®

Is the brand name of polymethacrylate-based copolymers. Eudragite is an amorphous polymer having a transition temperature between 9 and more than 150 0C. This polymer is non-biodegradable, non-toxic and non-absorbable [65]. Eudragite EPO with the active substance osthole has a solubility parameter of 20.55 MPa^{1/2}, $\Delta\delta$ (the ratio of drug to polymer solubility) 2.76 and is classified as a miscible.

The comparison of osthole drugs with EPO 1: 6 and 1: 9 was successful in increasing dissolution and decreasing the crystallinity of the active substance and colloid dispersed in the carrier [12]. Eudragite is used as a polymer in the manufacture of mefenamic acid disintegration tablets for taste masking using the HME method [31].

Kollidon VA 64® / Copovidone

Is the brand name of vinylpyrrolidone-vinyl acetate copolymer which is soluble in water and alcohol. Kollidon VA64 glass temperature is 105 ° C and melting point 160 ° C, complex viscosity 10,000 to 1000 Pa.s. The extrusion temperature is chosen in the itraconazole experiment with a ratio of 1: 4: 1 w / w ITZ-glutaric acid-Kollidon® VA 64 was 95 °C [13].

Soluplus®

Is the brand name of polyethyleneglycol-polyvinyl caprolactam-polyvinyl acetate

grafted copolymer. With a ratio of PEG 6000: Vinilcaprolactam: vinyl acetate 13:57:30. It has a molecular weight of 118,000 g / mol [64]. The glass transition temperature is 70°C. The low Tg makes Soluplus® good in the extraction process of Carbamazepine which is relatively low in temperature because Soluplus® is fast, low hygroscopic and will help maintain stability during storage [42].

Soluplus® provides the best solubility 160 fold increase in water solubility at a ratio of 1: 9, namely with the active substance 10% w/w piperine / Soluplus extrudates and the dissolution reaches 95% [66]. The solid dispersion lacidipine with soluplus® polymer is more physically stable than the VA 64® kollidon polymer [23].

Poloxamer (PLX)

Poloxamers are polyoxyethylene, polyoxypropylene block polymers [67]. Poloxamer is commonly used as solubilizing agents. However, in research conducted by Hwee Jing Ong and Rodolfo Pinal in 2018 the PLX 338 functions as a wetting agent, not a solubilizer. PLX 338 shows that the 20% (w/w) concentration of PLX is suitable for extrusion by taking only 5 minutes out of the extruder [41]. Other studies suggest that the combination of soluplus and poloxamer (2,3: 1) increases the dissolution of meloxicam and aids stability [63].

Poly (Ethylene Oxide) (PEO)

Having a low melting point of 62 °C-69 °C, the molecular weight of PEO decreases significantly after going through the HME process. PEO is degraded between temperatures of 330 °C – 450 °C so it is suitable for use as HME polymers that use high energy. Comparison of the drug Bicalutamide with polymeric polymer 1:10 had significant drug release over the 60 min period. During the extrusion process, PEO forms a gel layer with changes in thickness, composition and structure so that it is well hydrated and dissolved [57].

HPMC (Hydroxypropyl Methylcellulose)

In the preparation of griseofulvin drug extrudates 10% and 20% with 90% and 20% HPMC, the extrudate results are transparent and have a degradation temperature above 200°C. HPMC has several grades including

E5, E15, and E50. HPMC grade E5 has a Tg of 170-180 °C, with a molecular weight of 28,700 g / mol; HPMC grade E 15 has Tg 170-180 °C, with a molecular weight of 60,300 g / mol; HPMC grade E 50 has a Tg of 170-180 °C, with a molecular weight of 86,700 g / mol. The hydrogen bonding of the polymer to the drug allows molecular stabilization and prevents stabilization [68].

PEG (Polyethylene Glycol)

The PEG used by HME was various, namely PEG 400, PEG 4000, PEG 6000. PEG 400 with a concentration of 10-20% was successful in making control release preparations on diclofenac sodium using the HME method [53]. The melting point of PEG 6000 is 57°C, a combination of the active substance quercetin and polymer PEG 6000 with the HME method that has been done, namely the ratio of 1: 7 [46].

Plasdone S-630

Plasdone® S-630 is a copolymer of vinylpyrrolidone and vinyl acetate copovidone with a ratio of 60/40, having an average molecular weight of 24,000 to 30,000. This polymer was developed as a filler in the manufacture of tablets by direct compression and yields better dissolution tests than other binders [69]. The DSC results illustrated the absence of crystalline peaks in plasdone S-630 polymer with a ratio of 10% and 90% bifendate drug Plasdone ® S-630 [5].

PVP K30

The use of PVP 30 was reported in the study of atovaquone 20% with PVP K30 70%. PVP K30 was chosen because it has good miscibility with atovaquone, with Tg PVP K30 = 172.81 °C [49]. PVP K30 can also be used as a gelling agent in the manufacture of enteric coatings [70].

Partially Hydrolyzed Polyvinyl Alcohol (PVOH)

PVOP has the brand name Parateck® MXP, with Tg 60.8 ° C, Tm PVOH 150-200 ° C. The solubility parameter was 32.52 MPa^{1/2}, Δδ (ratio of drug and polymer solubility) 8, 90. The miscibility between Ibrutinib (IBR) and each polymer is HPMCAS > PVPVA > PEG > Soluplus > PVP > HPMC > HPC > PVOH., PVOH has a low solubility capacity in dissolving IBR and producing solid solutions [50].

Kollidon®12 PF (PVP / polyvinyl pyrrolidone)

This polymer is used in the preparation of solid dispersion (Aripiprazole) with the addition of succinic acid as an acidifier [27], and as a polymer in the manufacture of lansoprazole enteric-coated tablets [71]. Tadalafil solid dispersion with Kollidon® 12 PF polymer shows a faster dissolution rate than Kollidon® VA 64 dissolution. Tadalafil has separated from Kollidon® 12 PF solid preparations due to a combination of erosion and diffusion mechanisms [72]. Each polymer has individual results that vary depending on the interaction of the active substance and the polymer.

The mechanism of solid dispersion using the HME method

The mechanism of the solid dispersion method with HME is in principle the same as for other solid dispersions. Polymers that are hydrophilic and some even have amphiphilic properties, such as soluplus in liquid media will form micelles that can trap hydrophobic drugs into the hydrophobic nucleus to increase solubility [73]. This HME is used to disperse the insoluble drug into the hydrophilic matrix at the molecular level including changes in the drug in the amorphous, crystalline, or between amorphous and crystalline form [74]. The following is illustrated using Figure 1, namely the process that occurs in the TSE tool.

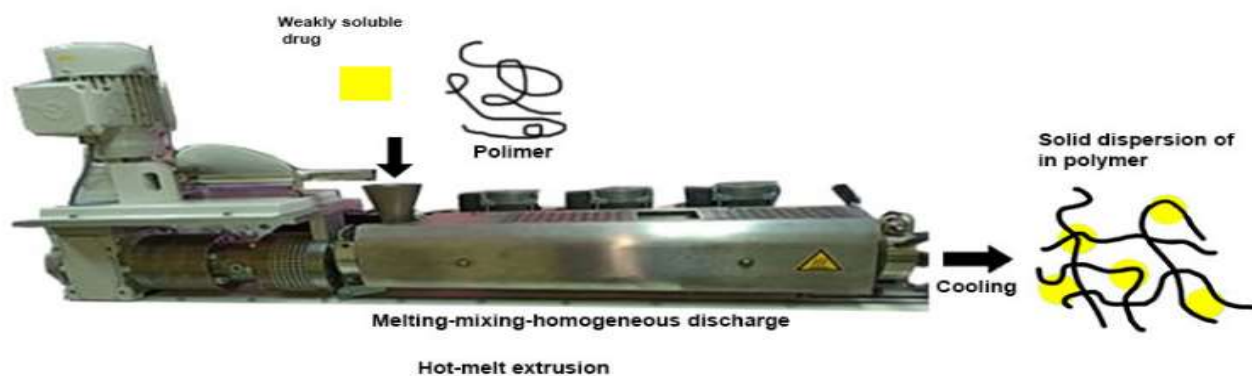


Fig 1: Illustration of HME on the Twin Screw Extruder(12)

Steps to Prepare Solid Dispersion by HME Method

Active substances that are usually processed through HME are those that have poor solubility in water because the advantages of this HME are that they can increase the solubility, dissolution rate, and bioavailability of active substances which

have limitations [75]. Solubility is always a major problem in product manufacturing, solubility is closely related to the melting point and partition coefficient. Solubility is related to the crystallinity and interactions between the solute and the solvent. General equity Equation (GSE) proposed Yalkowsky [76]:

$$\text{Log } S_w = 0,5 - 0,01 (\text{MP} - 25) - \text{log } K_{ow}$$

S_w = Solubility in water (molar)

MP = melting point

K_{ow} = the octanol–water partition coefficient of the solute.

If the marine substance has a melting point below 25 °C, if it is liquid then the MP-5 temperature is set to zero. In this review, the steps for preparing solid dispersions will be explained, namely: We mix the active substance with a hydrophilic carrier with the desired ratio (drug: polymer, w/w) to obtain a physical mixture (PM) then put it in a

powder mixer for about 20 minutes so that it is homogeneous before being put into the extruder machine. The final step is to enter API, polymer if necessary; you can add plasticizers, surfactants or others without adding solvents. By itself, we will get the results in the form of pellets according to the die that we set or want [28] [34].

This solid dispersion is usually unstable during storage; with this HME process will increase the stability of the active substance. If the preparation is unstable, the dissolution rate and bioavailability will decrease due to the separation phase, in which the separation phase is a direct factor of instability towards crystallinity [23].

Determinants of Successful Solid Dispersion with HME

The HME method will be successful, of course, with good preparation that must be met or also called a critical factor. The following will discuss the stages of the solid dispersion approach with HME:

Pshychochemical Evaluation [48]

We must know the characteristics of the material to be processed, namely the API and the polymer, ranging from polymers (Tg, Tm, Td, solubility in solvents, including donors or double bond recipients), and drugs (Tg, Tm, solubility in solvents, including donors or double bond recipient, log P, chemical stability).

Thermodynamic Assessment [48]

In this thermodynamic assessment, theoretically, we must know the solubility parameter δ , the prediction of Tg, and the interaction parameter (γ). Meanwhile, for the experiment, we can try to find the Tg of amorphous dispersion using the DSC method.

The process when making the extrudate

The tool used in HME is a screw extruder; it can use a single screw extruder (SSE), or a twin-screw extruder (TSE). TSE is commonly used and produces excellent mixing rates. We have to optimize the temperature, screw rotation, torque, die temperature, shear-stress time, residence time. Extraction temperature concerning Tg in both the polymer and the active substance. To obtain the desired product from both maximum mixing and flowability, the extraction temperature is usually set to 30-60 ° C above the Tg or Tm of the polymer [77,78].

Statistical Testing [48]

By looking at the correlation between chemical physics and thermodynamics.

Test HME [48]

3 things related to testing HME, namely:

- Solid dispersion characterization includes FTIR, SEM, XRD, DSC, TGA.
- Chemical interaction analysis (Raman spectroscopy)
- Pharmaceutical characterization (dissolution test, process evaluation)

Characterization of HME Solid Dispersion

To evaluate the results of the solid dispersion of HME applications, an examination can be done in the form of:

- Differential scanning calorimetry (DSC), DSC analysis is used as a first step to detect phase changes in the active substance and polymer used, as well as the phase transformations of the solid dispersion results. The results of this DSC examination, we get data on melting, miscibility, glass temperature, recrystallization from the solid dispersion results, and also to identify the presence of amorphous structures [79, 80].
- PXRD (Power X-ray diffraction), is used to see the crystallinity of preparation, in this case, the solid dispersion result [80]. If no crystalline peaks appear on the diffractogram, it indicates that the solid HME dispersion is amorphous and the amorphous form has high solubility compared to the crystalline form.
- TGA (Thermogravimetric analysis) to determine the degradation profile of drugs [81] and to see the thermal stability of single active substances or with polymers [28]. The TGA test gives us information that at what temperature the drug is degraded, thus assuring us that the temperature used during the melting process using the HME method does not damage the active substance of the preparation. It is hoped that the active substance can still show its action as a treatment because it is not degraded. The extrusion temperature greatly affects drug degradation, so selecting a low temperature on the TSE device is a prerequisite for obtaining high dissolution results [5].
- FT-IR (Fourier-Transform Infra-Red), this analysis is to see the bond between the active substance and the polymer. After HME treatment, did the peak position shift

or did a loss or even a new transmitted peak appears. HME is successful if no new transmittance peaks are found, but only a shift in the spectrum that can be explained by the amorphous nature of the HME results [73].

- SEM (Scanning Electron Microscopy), to see the crystalline properties of HME, see the morphology and particle size of HME results [82]. The success of HME is if the crystalline form of the active substance is no longer visible, but the API has already been dispersed into the polymer.

Conclusion

HME is a promising method and has been successful with a variety of drug delivery systems such as tablets, control releases, nano lipids, transdermals, nanoemulsions and others. The purpose of various HME can be to increase the solubility, dissolution rate and bioavailability of API (Active Pharmaceutical Ingredient) which has poor solubility in water, as well as taste masking. The selection of polymer and API is the key to the success of this method; we must study the physical and chemical properties of the active substance and its polymer. HME extraction temperature is usually 30-60 °C above the polymer glass transition temperature. The choice of API must be resistant to heating because this HME uses a high energy melting method. The mechanism in HME is by amorphizing the active substance so that the API will be dispersed in the polymer matrix and will increase the solubility of the API.

References

1. Patil H, Tiwari R V, Repka MA (2016) Hot-Melt Extrusion: from Theory to Application in Pharmaceutical Formulation. AAPS Pharm Sci. Tech.,.
2. Solanki NG, Kathawala M, Serajuddin ATM (2019) Effects of Surfactants on Itraconazole-Hydroxypropyl Methylcellulose Acetate Succinate Solid Dispersion Prepared by Hot Melt Extrusion III: Tableting of Extrudates and Drug Release From Tablets. J. Pharm Sci.,
3. Haser A, Cao T, Lubach J, Listro T, Acquarulo L, Zhang F (2017) Melt extrusion vs. spray drying: The effect of processing methods on crystalline content of naproxen-povidone formulations. Eur. J.

Pharm Sci.

4. Hwang I, Kang CY, Park JB (2017) Advances in hot-melt extrusion technology toward pharmaceutical objectives. Journal of Pharmaceutical Investigation.
5. Feng J, Xu L, Gao R, Luo Y, Tang X (2012) Evaluation of polymer carriers with regard to the bioavailability enhancement of bifendate solid dispersions prepared by hot-melt extrusion. Drug Dev. Ind. Pharm.
6. Wilson M, Williams MA, Jones DS, Andrews GP (2012) Hot-melt extrusion technology and pharmaceutical application. Therapeutic Delivery.
7. Sarode AL, Wang P, Obara S, Worthen DR (2014) Supersaturation, nucleation, and crystal growth during single- and biphasic dissolution of amorphous solid dispersions: Polymer effects and implications for oral bioavailability enhancement of poorly water soluble drugs. Eur. J. Pharm Biopharm.
8. Jelena D, Ioannis N, Svetlana I, Zorica D, Kyriakos K (2013) Preparation of Carbamazepine-Soluplus(R) solid dispersions by Hot-Melt Extrusion, and prediction of drug-polymer miscibility by thermodynamic model fitting. Eur J. Pharm. Biopharm [Internet]. Available from: <http://www.ncbi.nlm.nih.gov/pubmed/23333900>
9. Alshafiee M, Aljammal MK, Markl D, Ward A, Walton K, Blunt L, et al (2019) Hot-melt extrusion process impact on polymer choice of glyburide solid dispersions: The effect of wettability and dissolution. Int. J. Pharm.
10. Bochmann ES, Üstüner EE, Gryczke A, Wagner KG (2017) Predicting melt rheology for hot-melt extrusion by means of a simple Tg-measurement. Eur. J. Pharm. Biopharm.
11. Haser A, Cao T, Lubach JW, Zhang F (2018) In Situ Salt Formation during Melt Extrusion for Improved Chemical Stability and Dissolution Performance of a Meloxicam-Copovidone Amorphous Solid Dispersion. Mol. Pharm.
12. Yun F, Kang A, Shan J, Zhao X, Bi X, Li J, et al (2014) Preparation of osthole-polymer solid dispersions by hot-melt extrusion for dissolution and bioavailability enhancement. Int. J. Pharm.

13. Parikh T, Serajuddin ATM (2018) Development of Fast-Dissolving Amorphous Solid Dispersion of Itraconazole by Melt Extrusion of its Mixture with Weak Organic Carboxylic Acid and Polymer. *Pharm Res.*
14. Ibrahim M, Zhang J, Repka M, Chen R (2019) Characterization of the Solid Physical State of API and Its Distribution in Pharmaceutical Hot Melt Extrudates Using Terahertz Raman Imaging. *AAPS Pharm Sci. Tech.*
15. Bhardwaj V, Trasi NS, Zemlyanov DY, Taylor LS (2018) Surface area normalized dissolution to study differences in itraconazole-copovidone solid dispersions prepared by spray-drying and hot melt extrusion. *Int. J. Pharm.*
16. Shi NQ, Wang SR, Zhang Y, Huo JS, Wang LN, Cai JH, et al (2019) Hot melt extrusion technology for improved dissolution, solubility and “spring-parachute” processes of amorphous self-micellizing solid dispersions containing BCS II drugs indomethacin and fenofibrate: Profiles and mechanisms. *Eur. J. Pharm. Sci.*
17. Ogawa N, Hiramatsu T, Suzuki R, Okamoto R, Shibagaki K, Fujita K, et al (2018) Improvement in the water solubility of drugs with a solid dispersion system by spray drying and hot-melt extrusion with using the amphiphilic polyvinyl caprolactam-polyvinyl acetate-polyethylene glycol graft copolymer and D-mannitol. *Eur. J. Pharm. Sci.*
18. Wen T, Niu B, Wu Q, Zhou Y, Pan X, Quan G, et al (2019) Fenofibrate Solid Dispersion Processed by Hot-Melt Extrusion: Elevated Bioavailability and Its Cell Transport Mechanism. *Curr Drug Deliv.*
19. Fule R, Dhamecha D, Maniruzzaman M, Khale A, Amin P (2015) Development of hot melt co-formulated antimalarial solid dispersion system in fixed dose form (ARLUMELT): Evaluating amorphous state and in vivo performance. *Int. J. Pharm.*
20. Benjasirimongkol P, Ueda K, Higashi K, Sriamornsak P, Moribe K (2018) An insight into stabilization mechanism of a solid dispersion of indomethacin/partially hydrolyzed polyvinyl alcohol prepared by hot-melt extrusion. *Chem Pharm Bull.*
21. Fan W, Zhang X, Zhu W, Di L (2020) The Preparation of Curcumin Sustained-Release Solid Dispersion by Hot-Melt Extrusion-II. Optimization of Preparation Process and Evaluation In Vitro and In Vivo. *J. Pharm. Sci.*
22. Chowdhury N, Vhora I, Patel K, Bagde A, Kutlehria S, Singh M (2018) Development of Hot Melt Extruded Solid Dispersion of Tamoxifen Citrate and Resveratrol for Synergistic Effects on Breast Cancer Cells. *AAPS Pharm Sci. Tech.*
23. Xi L, Song H, Wang Y, Gao H, Fu Q (2018) Lacidipine Amorphous Solid Dispersion Based on Hot Melt Extrusion: Good Miscibility, Enhanced Dissolution, and Favorable Stability. *AAPS Pharm Sci. Tech.*
24. Solanki NG, Gumaste SG, Shah A V, Serajuddin ATM (2019) Effects of Surfactants on Itraconazole-Hydroxypropyl Methylcellulose Acetate Succinate Solid Dispersion Prepared by Hot Melt Extrusion. II: Rheological Analysis and Extrudability Testing. *J. Pharm. Sci.*
25. Enose AA, Dasan PK, Sivaramakrishnan H, Shah SM (2014) Formulation and Characterization of Solid Dispersion Prepared by Hot Melt Mixing: A Fast Screening Approach for Polymer Selection. *J. Pharm.*
26. Kulkarni C, Kelly AL, Gough T, Jadhav V, Singh KK, Paradkar A (2018) Application of hot melt extrusion for improving bioavailability of artemisinin a thermolabile drug. *Drug Dev Ind Pharm.*
27. McFall H, Sarabu S, Shankar V, Bandari S, Murthy SN, Kolter K, et al (2019) Formulation of aripiprazole-loaded pH-modulated solid dispersions via hot-melt extrusion technology: In vitro and in vivo studies. *Int. J. Pharm.*
28. Ponnammal P, Kanaujia P, Yani Y, Ng WK, Tan RBH (2018) Orally disintegrating tablets containing melt extruded amorphous solid dispersion of tacrolimus for dissolution enhancement. *Pharmaceutics.*
29. Vo AQ, Feng X, Pimparade M, Ye X, Kim DW, Martin ST, et al (2017) Dual-mechanism gastroretentive drug delivery system loaded with an amorphous solid dispersion prepared by hot-melt extrusion.

- Eur. J. Pharm. Sci.
30. Walsh D, Serrano DR, Worku ZA, Madi AM, O'Connell P, Twamley B, et al (2018) Engineering of pharmaceutical cocrystals in an excipient matrix: Spray drying versus hot melt extrusion. *Int. J. Pharm.*
 31. Alshehri SM, Park JB, Alsulays BB, Tiwari R V, Almutairy B, Alshetaili AS, et al (2015) Mefenamic acid taste-masked oral disintegrating tablets with enhanced solubility via molecular interaction produced by hot melt extrusion technology. *J. Drug Deliv. Sci. Technol.*
 32. Lee JY, Kang WS, Piao J, Yoon IS, Kim DD, Cho HJ (2015) Soluplus®/TPGSGS-based solid dispersions prepared by hot-melt extrusion equipped with twin-screw systems for enhancing oral bioavailability of valsartan. *Drug Des Devel Ther.*
 33. Huang S, O'Donnell KP, Delpon de Vaux SM, O'Brien J, Stutzman J, Williams RO (2017) Processing thermally labile drugs by hot-melt extrusion: The lesson with gliclazide. *Eur. J. Pharm. Biopharm.*
 34. Gao N, Guo M, Fu Q, He Z (2017) Application of hot melt extrusion to enhance the dissolution and oral bioavailability of oleanolic acid. *Asian J. Pharm Sci.*
 35. Pawar J, Suryawanshi D, Moravkar K, Aware R, Shetty V, Maniruzzaman M, et al (2018) Study the influence of formulation process parameters on solubility and dissolution enhancement of efavirenz solid solutions prepared by hot-melt extrusion: a QbD methodology. *Drug Deliv Transl Res.*
 36. Lu J, Cuellar K, Hammer NI, Jo S, Gryczke A, Kolter K, et al (2016) Solid-state characterization of Felodipine-Soluplus amorphous solid dispersions. *Drug Dev Ind Pharm.*
 37. Alhijaj M, Bouman J, Wellner N, Belton P, Qi S (2015) Creating Drug Solubilization Compartments via Phase Separation in Multicomponent Buccal Patches Prepared by Direct Hot Melt Extrusion-Injection Molding. *Mol Pharm.*
 38. Liu X, Zhou L, Zhang F (2017) Reactive Melt Extrusion To Improve the Dissolution Performance and Physical Stability of Naproxen Amorphous Solid Dispersions. *Mol. Pharm.*
 39. Albarahmieh E, Qi S, Craig DQM (2016) Hot melt extruded transdermal films based on amorphous solid dispersions in Eudragit RS PO: The inclusion of hydrophilic additives to develop moisture-activated release systems. *Int. J. Pharm.*
 40. Martinez-Marcos L, Lamprou DA, McBurney RT, Halbert GW (2016) A novel hot-melt extrusion formulation of albendazole for increasing dissolution properties. *Int. J. Pharm.*
 41. Ong HJ, Pinal R (2018) Drug Solubilization by Means of a Surface-Modified Edible Biopolymer Enabled by Hot Melt Extrusion. *J. Pharm Sci.*
 42. Alshahrani SM, Lu W, Park JB, Morott JT, Alsulays BB, Majumdar S, et al (2015) Stability-enhanced Hot-melt Extruded Amorphous Solid Dispersions via Combinations of Soluplus® and HPMCAS-HF. *AAPS Pharm Sci. Tech.*
 43. Alshehri SM, Tiwari R V, Alsulays BB, Ashour EA, Alshetaili AS, Almutairy B, et al (2017) Investigation of the combined effect of MgO and PEG on the release profile of mefenamic acid prepared via hot-melt extrusion techniques. *Pharm Dev Technol.*
 44. Koo JS, Lee SY, Nam S, Azad MOK, Kim M, Kim K, et al (2018) Preparation of cupric sulfate-based self-emulsifiable nanocomposites and their application to the photothermal therapy of colon adenocarcinoma. *Biochem Biophys Res Commun.*
 45. Auch C, Harms M, Mäder K (2019) How changes in molecular weight and PDI of a polymer in amorphous solid dispersions impact dissolution performance. *Int. J. Pharm.*
 46. Shi X, Fan N, Zhang G, Sun J, He Z, Li J (2020) Quercetin amorphous solid dispersions prepared by hot melt extrusion with enhanced solubility and intestinal absorption. *Pharm. Dev. Technol.*
 47. Duque L, Körber M, Bodmeier R (2018) Impact of change of matrix crystallinity and polymorphism on ovalbumin release from lipid-based implants. *Eur J. Pharm Sci.*
 48. Simões MF, Pereira A, Cardoso S, Cadonau S, Werner K, Pinto RMA, et al (2020) Five-Stage Approach for a

- Systematic Screening and Development of Etravirine Amorphous Solid Dispersions by Hot-Melt Extrusion. *Mol. Pharm.*
49. Takabe H, Warnken ZN, Zhang Y, Davis DA, Smyth HDC, Kuhn JG, et al (2018) A repurposed drug for brain cancer: Enhanced atovaquone amorphous solid dispersion by combining a spontaneously emulsifying component with a polymer carrier. *Pharmaceutics*.
 50. Simões MF, Nogueira BA, Tabanez AM, Fausto R, Pinto RMA, Simões S (2020) Enhanced solid-state stability of amorphous ibrutinib formulations prepared by hot-melt extrusion. *Int. J. Pharm.*
 51. Abu-Diak OA, Jones DS, Andrews GP (2011) An investigation into the dissolution properties of celecoxib melt extrudates: Understanding the role of polymer type and concentration in stabilizing supersaturated drug concentrations. *Mol. Pharm.*
 52. Hwang I, Renuka V, Lee JH, Weon KY, Kang CY, Lee BJ, et al (2020) Preparation of celecoxib tablet by hot melt extrusion technology and application of process analysis technology to discriminate solubilization effect. *Pharm. Dev. Technol.*
 53. Bisharat L, Alkhatib HS, Abdelhafez A, Barqawi A, Aljaberi A, Qi S, et al (2020) Hot melt extruded zein for controlled delivery of diclofenac sodium: Effect of drug loading and medium composition. *Int. J. Pharm.*
 54. Dong L, Mai Y, Liu Q, Zhang W, Yang J (2020) Mechanism and improved dissolution of glycyrrhetic acid solid dispersion by alkalizers. *Pharmaceutics*.
 55. Knapik-Kowalczyk J, Chmiel K, Jurkiewicz K, Wojnarowska Z, Kurek M, Jachowicz R, et al (2019) Influence of Polymeric Additive on the Physical Stability and Viscoelastic Properties of Aripiprazole. *Mol. Pharm.*
 56. Solanki NG, Tahsin M, Shah A V, Serajuddin ATM (2018) Formulation of 3D Printed Tablet for Rapid Drug Release by Fused Deposition Modeling: Screening Polymers for Drug Release, Drug-Polymer Miscibility and Printability. *J. Pharm. Sci.*
 57. Abu-Diak OA, Jones DS, Andrews GP (2012) Understanding the performance of melt-extruded poly(ethylene oxide)-bicalutamide solid dispersions: Characterisation of microstructural properties using thermal, spectroscopic and drug release methods. *J. Pharm. Sci.*
 58. Tres F, Coombes SR, Phillips AR, Hughes LP, Wren SAC, Aylott JW, et al (2015) Investigating the dissolution performance of amorphous solid dispersions using magnetic resonance imaging and proton NMR. *Molecules*.
 59. Tres F, Treacher K, Booth J, Hughes LP, Wren SAC, Aylott JW, et al (2016) Indomethacin-Kollidon VA64 Extrudates: A Mechanistic Study of pH-Dependent Controlled Release. *Mol. Pharm.*
 60. Albarahmeh E, Alkhalidi BA, Al-Hiari Y (2020) Evaluation of amorphous dispersion of a cellulose ester-colophony mix for ibuprofen controlled release processed by HME and spin coating. *Carbohydr Polym.*
 61. Moseson DE, Parker AS, Beaudoin SP, Taylor LS (2020) Amorphous solid dispersions containing residual crystallinity: Influence of seed properties and polymer adsorption on dissolution performance. *Eur. J. Pharm. Sci.*
 62. Ashour EA, Majumdar S, Alsheteli A, Alshehri S, Alsulays B, Feng X, et al (2016) Hot melt extrusion as an approach to improve solubility, permeability and oral absorption of a psychoactive natural product, piperine. *J. Pharm. Pharmacol.*
 63. Taha NF, Emam MF, Emara LH (2020) A novel combination of Soluplus®/Poloxamer for Meloxicam solid dispersions via hot melt extrusion for rapid onset of action. Part 2: comparative bioavailability and IVIVC. *Drug Dev. Ind. Pharm.*
 64. Kolter, Karl and Karl M (2010) Suitability of Plasticized Polymers for Hot Melt Extrusion. *Excipients&Actives For Pharma* [Internet]. 24: 2-6. Available from: www.pharma-ingredients.basf.com
 65. Thakral S, Thakral NK, Majumdar DK (2013) Eudragit®: A technology evaluation. *Expert Opinion on Drug Delivery*.
 66. Ashour EA, Majumdar S, Alsheteli A, Alshehri S, Alsulays B, Feng X, et al (2016) Hot melt extrusion as an approach to improve solubility, permeability and oral absorption of a psychoactive natural product, piperine. *J. Pharm. Pharmacol.*, 989-98.

67. McLain VC (2008) Safety assessment of poloxamers 101, 105, 108, 122, 123, 124, 181, 182, 183, 184, 185, 188, 212, 215, 217, 231, 234, 235, 237, 238, 282, 284, 288, 331, 333, 334, 335, 338, 401, 402, 403, and 407, poloxamer 105 benzoate, and poloxamer 182 dibenzoate as use. *International Journal of Toxicology*.
68. LaFontaine JS, Prasad LK, Brough C, Miller DA, McGinity JW, Williams RO (2016) Thermal Processing of PVP- and HPMC-Based Amorphous Solid Dispersions. *AAPS Pharm. Sci. Tech.*
69. Jayachandra Babu R, Brostow W, Kalogeras IM, Sathigari S (2009) Glass transitions in binary drug + polymer systems. *Mater Lett.*
70. Andrews GP, Jones DS, Diak OA, McCoy CP, Watts AB, McGinity JW (2008) The manufacture and characterisation of hot-melt extruded enteric tablets. *Eur. J. Pharm Biopharm.*
71. Alsulays BB, Kulkarni V, Alshehri SM, Almutairy BK, Ashour EA, Morott JT, et al (2017) Preparation and evaluation of enteric coated tablets of hot-melt extruded lansoprazole. *Drug Dev. Ind. Pharm.*
72. Školáková T, Slámová M, Školáková A, Kadeřábková A, Patera J, Zámostný P (2019) Investigation of dissolution mechanism and release kinetics of poorly water-soluble tadalafil from amorphous solid dispersions prepared by various methods. *Pharmaceutics*.
73. Soulairol I, Tarlier N, Bataille B, Cacciaguerra T, Sharkawi T (2015) Spray-dried solid dispersions of nifedipine and vinylcaprolactam/vinylacetate/PEG6000 for compacted oral formulations. *Int. J. Pharm.*
74. Zhang X, Wang M, Li P, Wang A, Liang R, Gai Y, et al (2016) Application of hot-melt extrusion technology for designing an elementary osmotic pump system combined with solid dispersion for a novel poorly water-soluble antidepressant. *Pharm. Dev. Technol.*
75. Patil H, Feng X, Ye X, Majumdar S, Repka MA (2015) Continuous Production of Fenofibrate Solid Lipid Nanoparticles by Hot-Melt Extrusion Technology: a Systematic Study Based on a Quality by Design Approach. *AAPS J.*
76. Ran Y, Yalkowsky SH (2001) Prediction of drug solubility by the general solubility equation (GSE). *J. Chem. Inf. Comput Sci.*
77. Chan SY, Qi S, Craig DQM (2015) An investigation into the influence of drug-polymer interactions on the miscibility, processability and structure of polyvinylpyrrolidone-based hot melt extrusion formulations. *Int. J. Pharm.*
78. Chokshi RJ, Sandhu HK, Iyer RM, Shah NH, Malick AW, Zia H (2005) Characterization of physico-mechanical properties of indomethacin and polymers to assess their suitability for hot-melt extrusion process as a means to manufacture solid dispersion/solution. *J. Pharm Sci.*
79. Shah S, Maddineni S, Lu J, Repka MA (2013) Melt extrusion with poorly soluble drugs. *International Journal of Pharmaceutics*.
80. Vasoya JM, Desai HH, Gumaste SG, Tillotson J, Kelemen D, Dalrymple DM, et al (2019) Development of Solid Dispersion by Hot Melt Extrusion Using Mixtures of Polyoxylglycerides With Polymers as Carriers for Increasing Dissolution Rate of a Poorly Soluble Drug Model. *J. Pharm. Sci.*
81. Goyanes A, Allahham N, Trenfield SJ, Stoyanov E, Gaisford S, Basit AW (2019) Direct powder extrusion 3D printing: Fabrication of drug products using a novel single-step process. *Int. J. Pharm.*
82. Qi S, Belton P, Nollenberger K, Clayden N, Reading M, Craig DQM (2010) Characterisation and prediction of phase separation in hot-melt extruded solid dispersions: A thermal, microscopic and NMR relaxometry study. *Pharm Res.*